



Accelerating the new hydrogen economy in the Midlands

Hydrogen Summer School

Day 3:

University of Birmingham

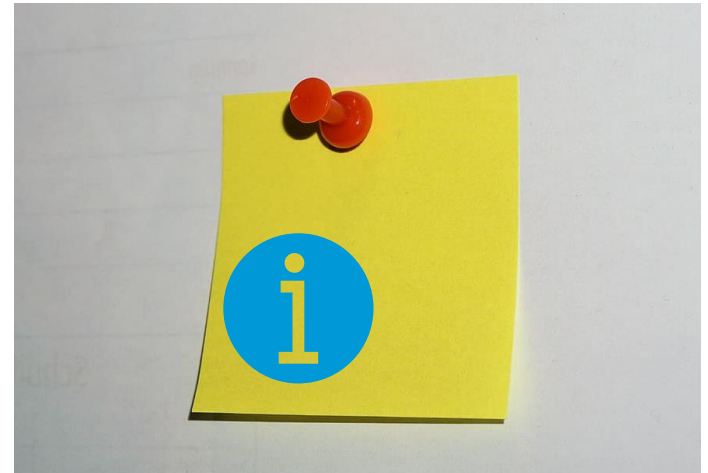
11th September 2024



Housekeeping



- Fire Alarm
- Fire Exits and Assembly Point
- Toilets
- Mobile phones
Please put on silent, if you need to take or make a call, please leave the room.
- Questions
You will have the opportunity to ask questions between presentations and throughout the day.



Agenda

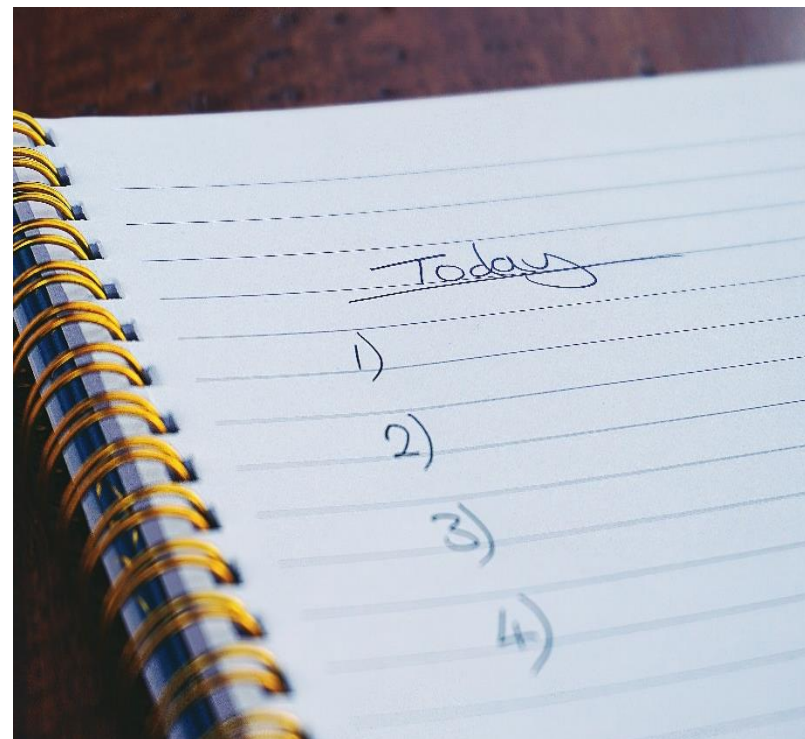


Morning:

- Welcome & Fuel cells for transport applications and hydrogen fuel – Dr Yousif Al-Sagheer
- Fuel cell seminars x2 – Dr Hadi Heidary, Dr Bernardo Sarruf, Dr Kun Zhang
- Hydrogen production via biomass conversion, Dr Paula Blanco-Sanchez (Aston University)
- Hydrogen-based fuels – Dr Artur Majewski

Afternoon:

- Tyseley Energy Park visit and tour
- Networking and Q&A





Welcome to University of Birmingham

- School of Chemical Engineering
- Fuel Cells and Hydrogen Research Centre
- Birmingham Energy Innovation Centre (BEIC)
- Key partner university in HyDEX Project
- Hydrogen demonstrator facilities at Tyseley Energy Park
- Research and innovation in waste, energy and low carbon vehicle systems



Dr Yousif Al-Sagheer



Fuel cells for transport applications

Fuel Cells and Hydrogen Research Centre

Fuel cells for transport applications

- Fuel cell principle
- Fuel cell system
- Hydrogen production shades
- Decarbonising transport
- Fuel cell vehicles
- Fuel cell hybrid EV
- Off-road Vehicle: Fuel cell tractor
- Fuel cell in maritime applications

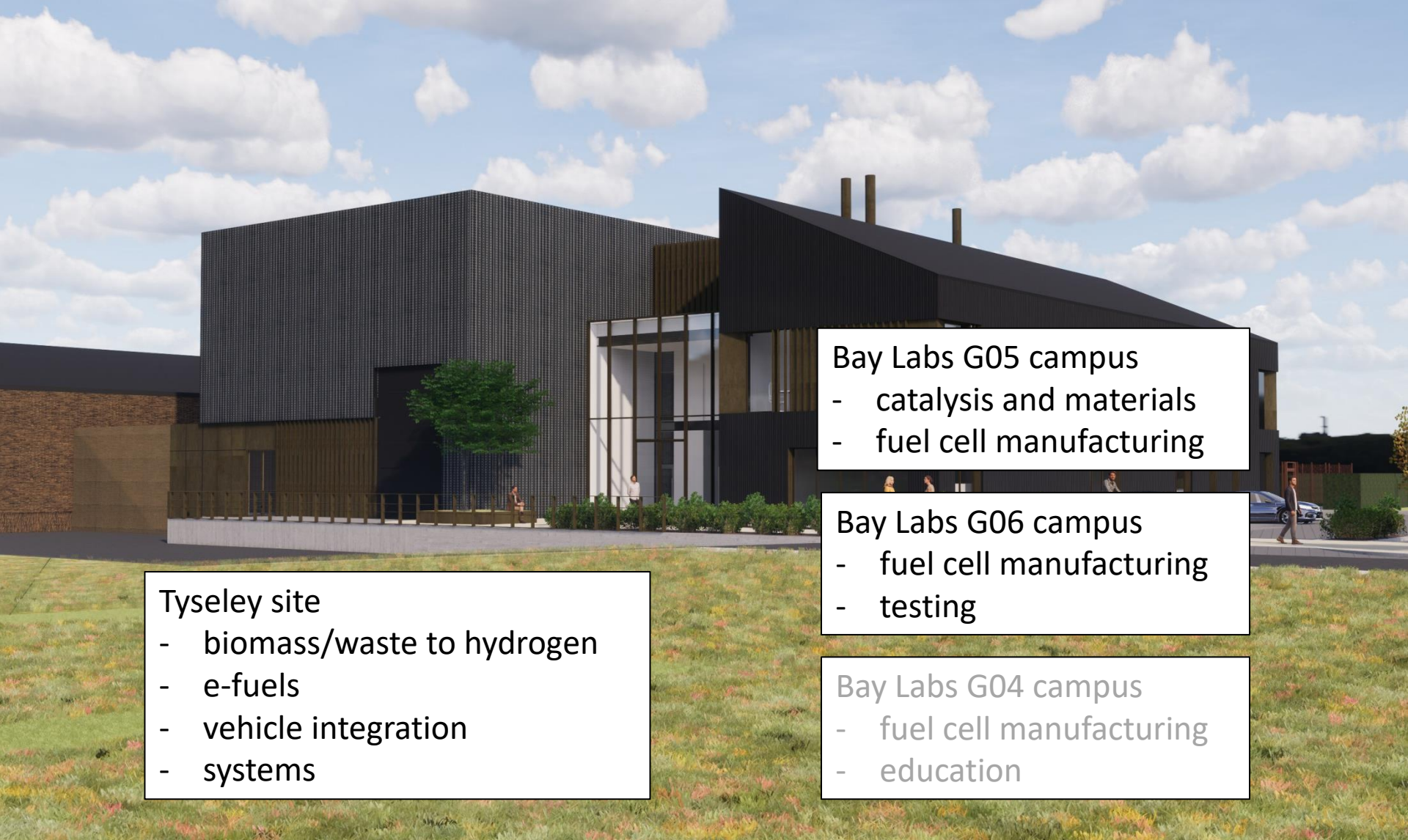
Portfolio Overview

A group of 14 staff and 40 PhD and MRes students working in:

- **Hydrogen and Synthetic Fuel Production**
- **Fuel production from biomass and waste**
- **Low Temperature Fuel Cells & Electrolysis**
- **High Temperature Fuel Cells & Electrolysis**
- **Socio-techno-economic studies**
- **Educational initiatives**

The primary application is:

- **Integration of fuel cell systems on vehicles**



Tyseley site

- biomass/waste to hydrogen
- e-fuels
- vehicle integration
- systems

Bay Labs G05 campus

- catalysis and materials
- fuel cell manufacturing

Bay Labs G06 campus

- fuel cell manufacturing
- testing

Bay Labs G04 campus

- fuel cell manufacturing
- education

Fuel cell principle / PEM Fuel Cells

Anode: $\text{H}_2 (\text{g}) \rightarrow 2\text{H}^+ (\text{aq}) + 2\text{e}^-$

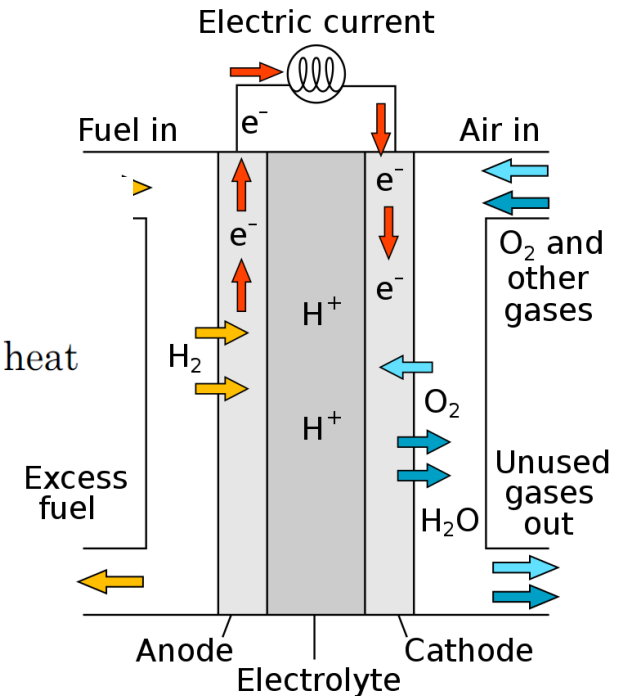
Cathode: $\frac{1}{2}\text{O}_2 (\text{g}) + 2\text{H}^+ (\text{aq}) + 2\text{e}^- \rightarrow \text{H}_2\text{O} (\text{l})$

Overall: $\text{H}_2 (\text{g}) + \frac{1}{2}\text{O}_2 (\text{g}) \rightarrow \text{H}_2\text{O} (\text{l}) + \text{electric energy} + \text{waste heat}$

$$\text{H}_2 \text{ usage} = \frac{In}{2F} \text{ moles s}^{-1}$$

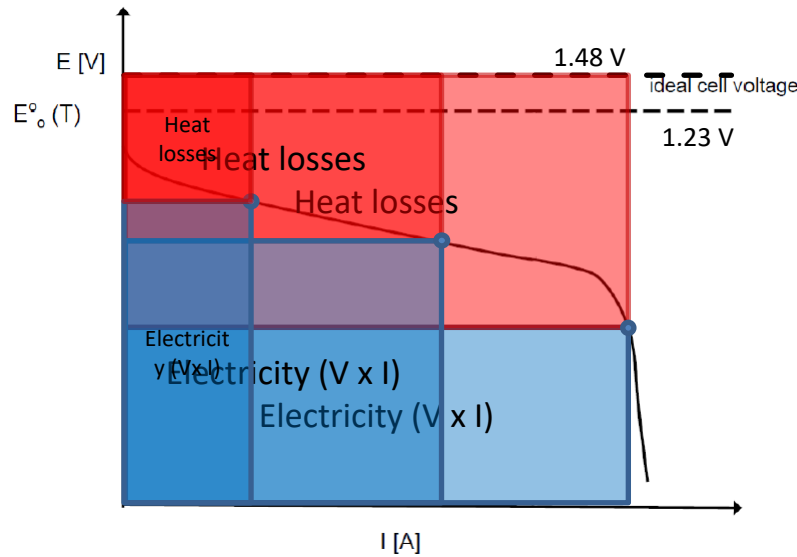
$$\text{O}_2 \text{ usage} = \frac{In}{4F} \text{ moles s}^{-1}$$

$$E = \frac{-\Delta \bar{g}_f}{2F}$$



Fuel cell polarisation / I-V curve

$$E = \frac{-\Delta \bar{g}_f}{2F}$$



$$\text{Efficiency} = \frac{\text{FC electrical power (V} \times \text{I)}}{\text{H2 enery input (E} \times \text{I)}}$$

$$\text{Efficiency} = \frac{V}{E}$$

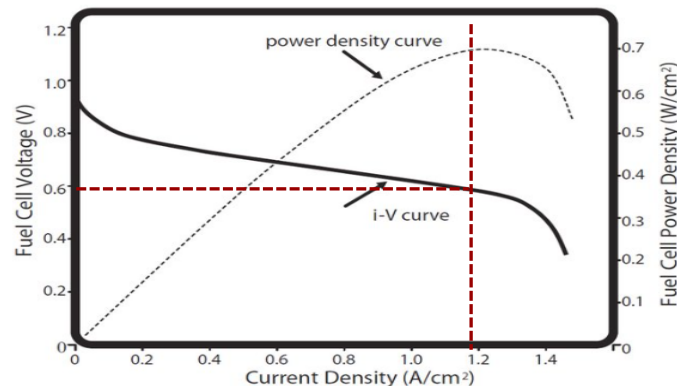
$$\text{Efficiency} = \frac{V}{1.23} \quad \text{w.r.t LHV}$$

if the water product was in vapour form

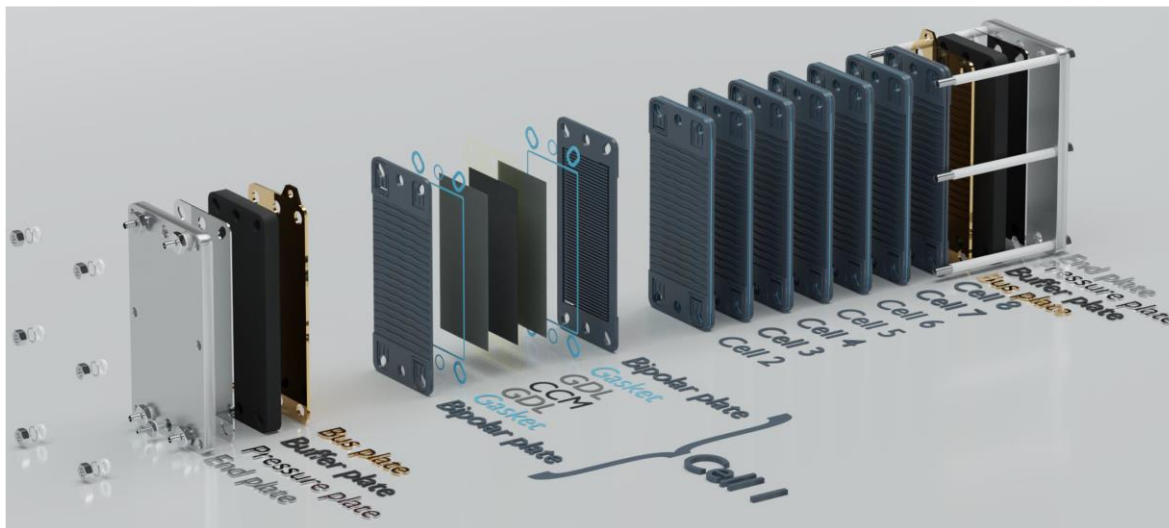
$$\text{Efficiency} = \frac{V}{1.48} \quad \text{w.r.t HHV}$$

if the water product was in liquid form

$$\text{Efficiency} = \frac{0.6}{1.23} = 49\% \quad \text{w.r.t LHV}$$



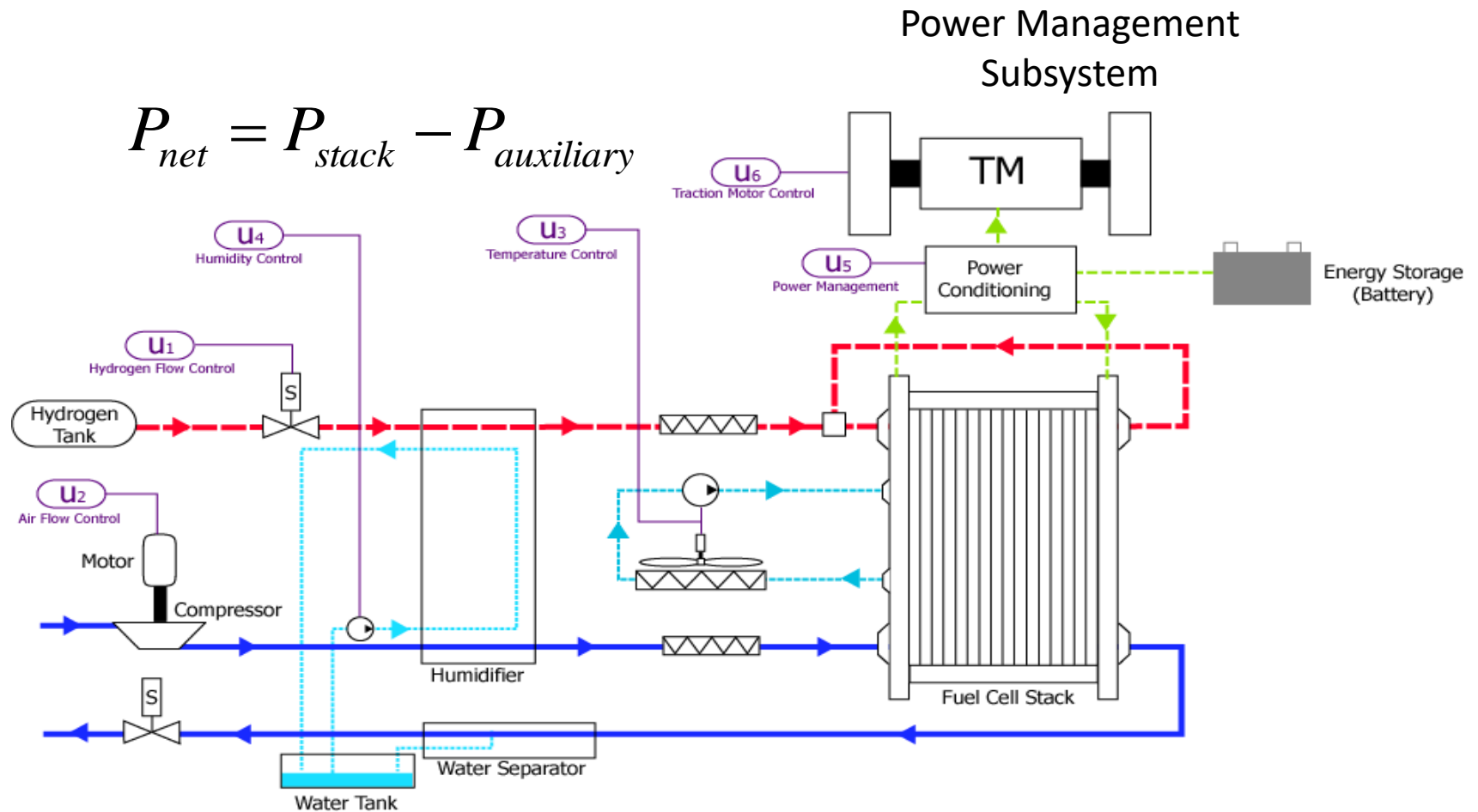
PEFC Stack Design



- Number of cells
 - nominal stack voltage
 - OCV voltage (max)
- Cell area
 - Current density
 - Stack current

Fuel Cell System

$$P_{net} = P_{stack} - P_{auxiliary}$$

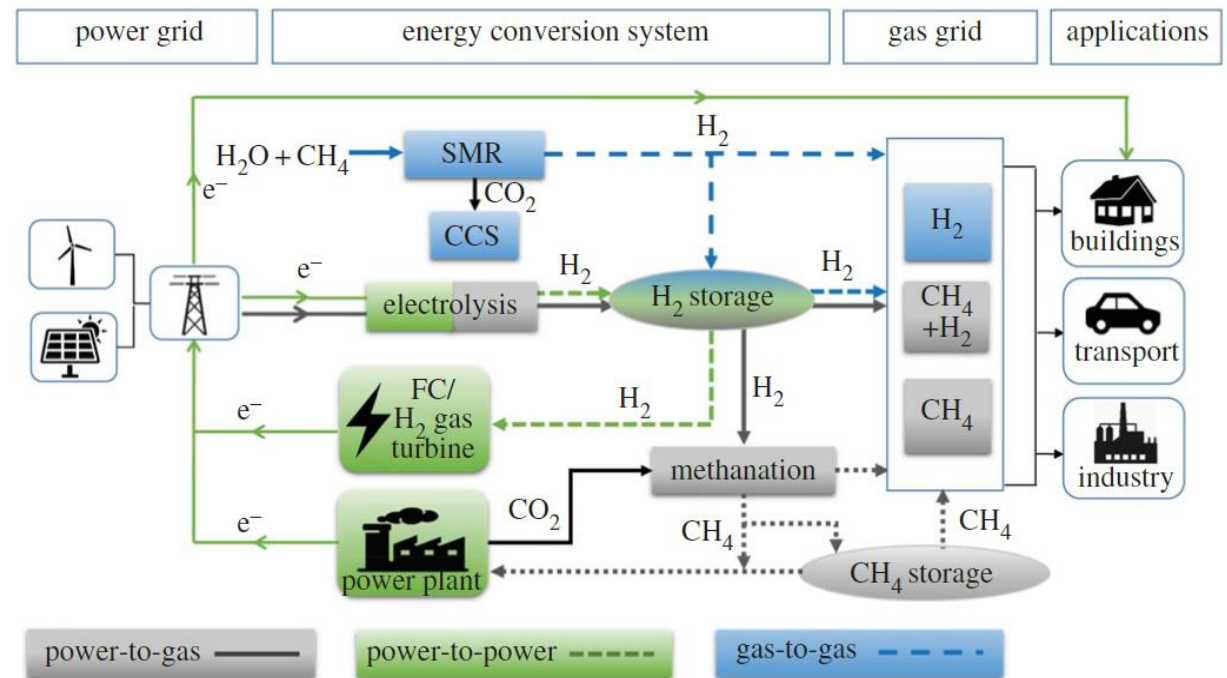


Source: Stephanopoulos

Why Hydrogen?

The role of H₂ in clean energy future

- Highest gravimetric energy density 33.6 kWh/kg (LHV)
- No emission at the point of use
- Issue: Low volumetric energy density
- Energy vector
- Interlink multi energy sectors



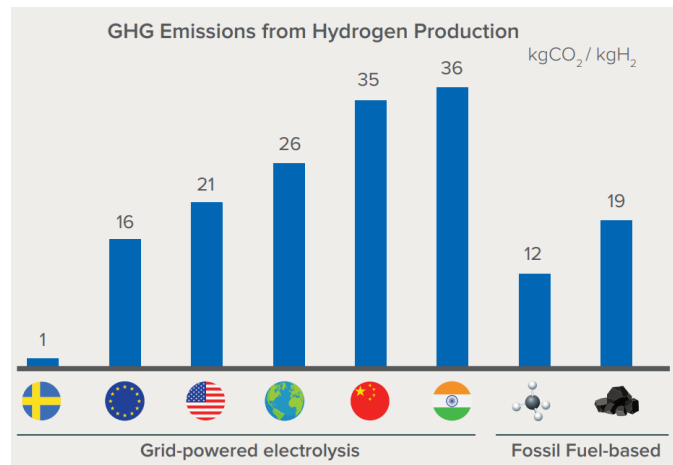
Clean energy and the hydrogen economy

Different shades of Hydrogen

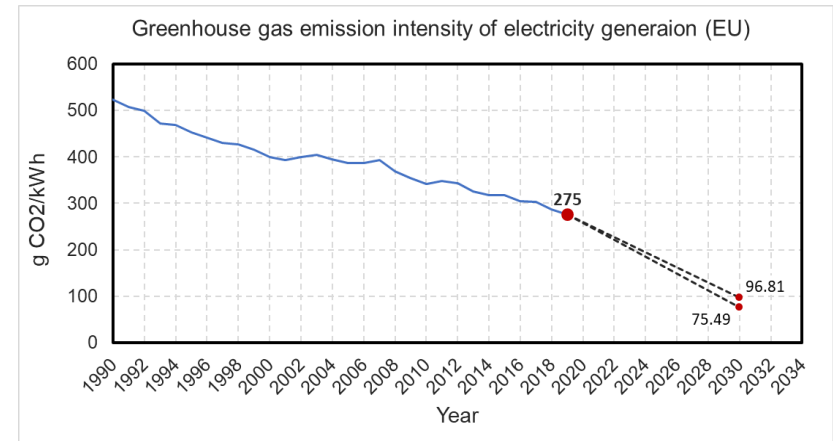
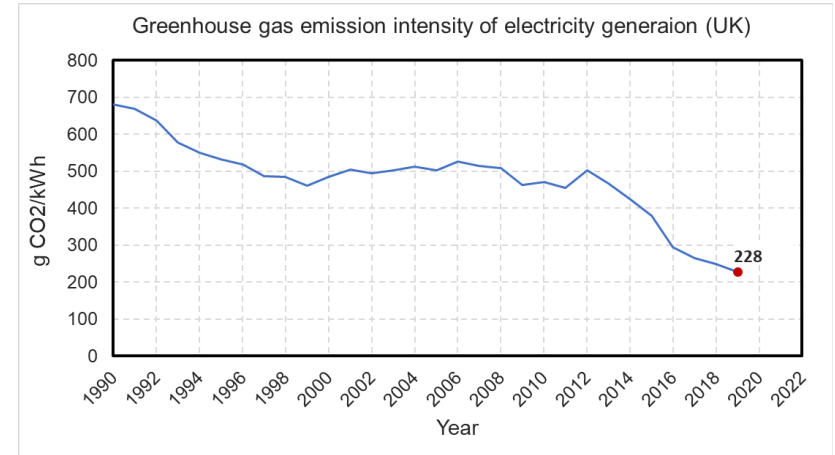
Colors	Black Hydrogen	Grey Hydrogen	Blue Hydrogen	Turquoise Hydrogen	Yellow Hydrogen	Pink Hydrogen	Green Hydrogen
Process	Gasification	SMR	SMR or gasification with carbon capture (85-95%)	Pyrolysis	Sulfur– iodine cycle Electrolysis	Electrolysis	Electrolysis
Source	Coal	Methane	Methane or coal	Methane	Nuclear power Grid electricity	Nuclear power	Renewable Energy
Brown Gasification fossil fuel			White H2 as a by-product of industrial processes OR H2 naturally occurring underground				

Carbon footprint per kWh (Grid electricity)

- Compare to Steam Methane Reforming (SMR)
 - SMR: 8-12 kg CO₂/ kg of H₂
- Electrolysis: 39.4 kWh/kg HHV with 70% eff.
 - ~ 56 kWh/kg H₂
- EU grid (average): ~15.4 kg CO₂/ kg of H₂
- UK grid: 12.8 kg CO₂/ kg of H₂



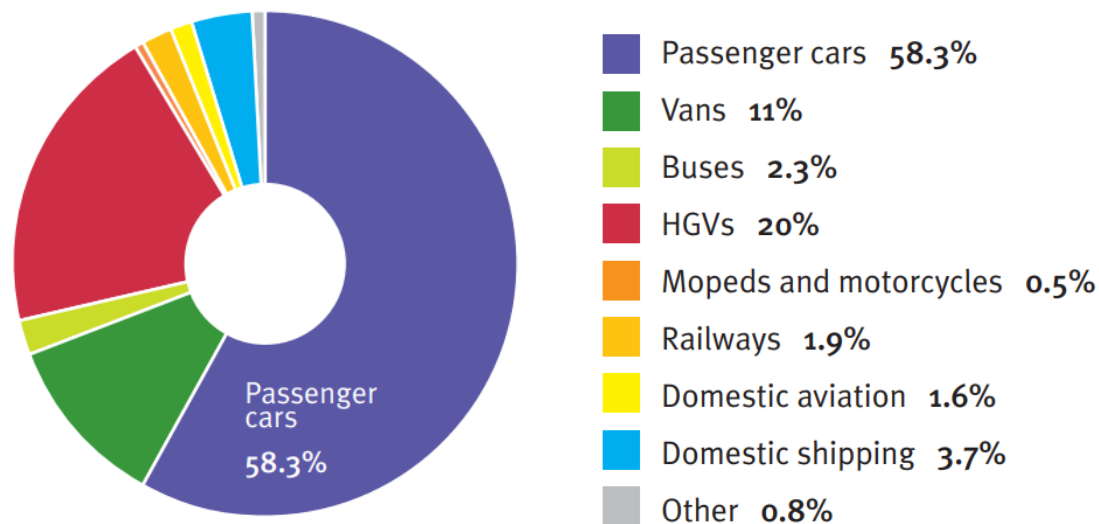
Hydrogen's Decarbonization Impact for Industry



European Environment Agency

Decarbonising Transport ?

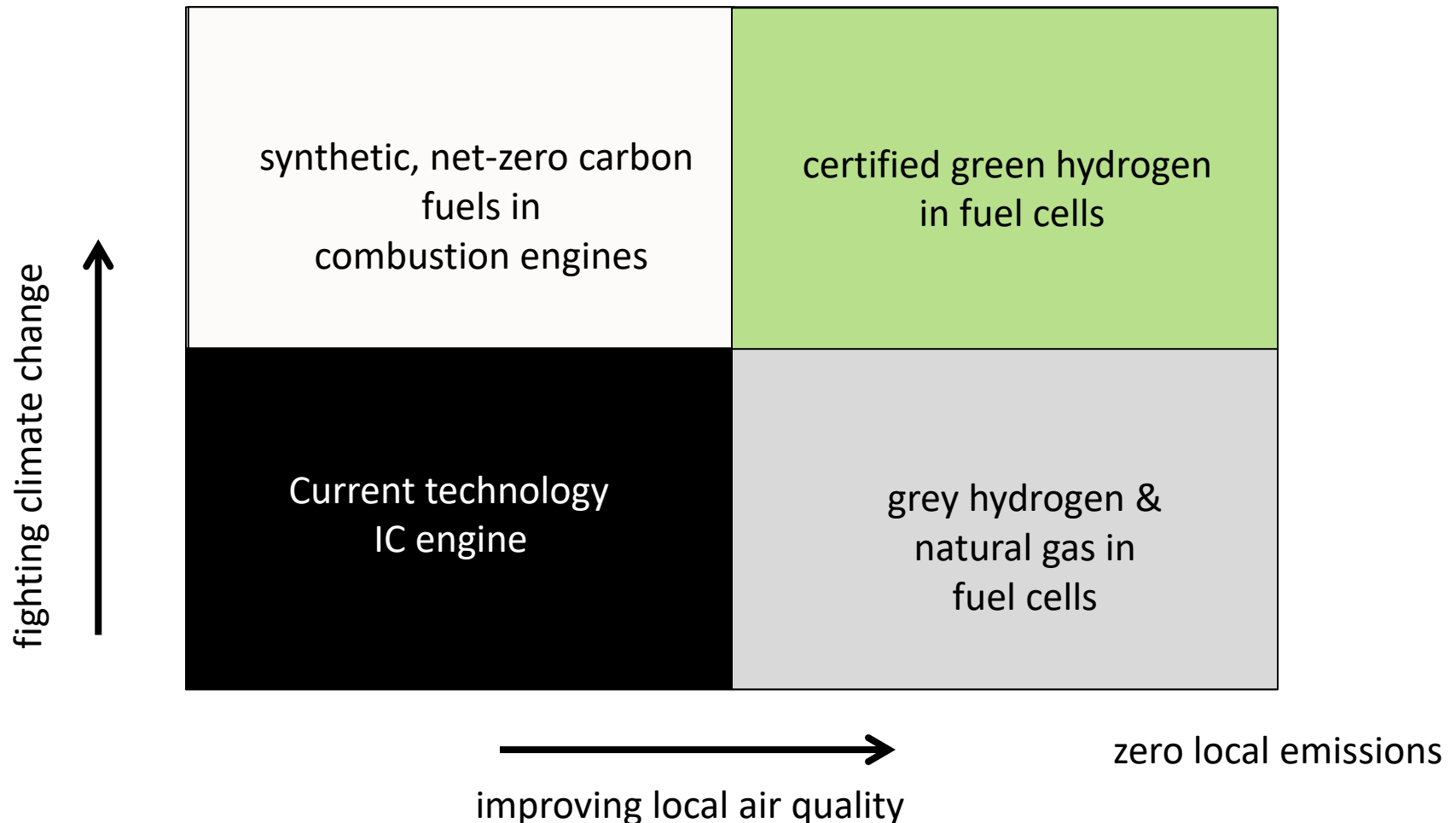
- Greenhouse Gas (GHG) emissions reduction
 - Emissions need to be reduced by 50-60% by 2050.
 - 27% of global CO₂ emissions are from the transport sector
 - For UK, 92% of total transport emissions is from domestic transport.
 - Globally, 73% of transport related CO₂ emissions are from road transport



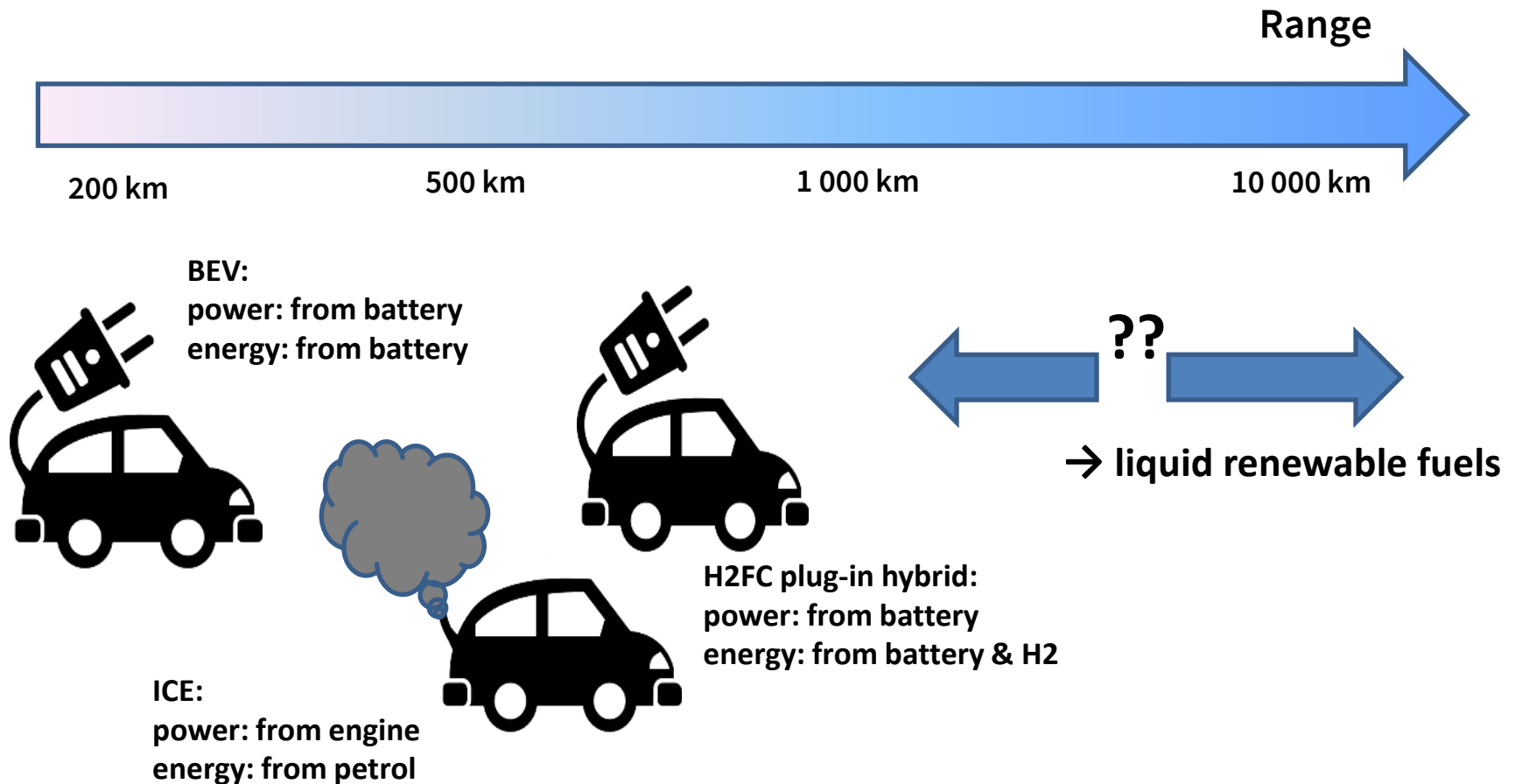
UK domestic transport GHG emissions 2009 excluding travel across borders / Grantham Institute

Global vs. Local Zero Emissions

Climate neutral



Electric Transport Solutions

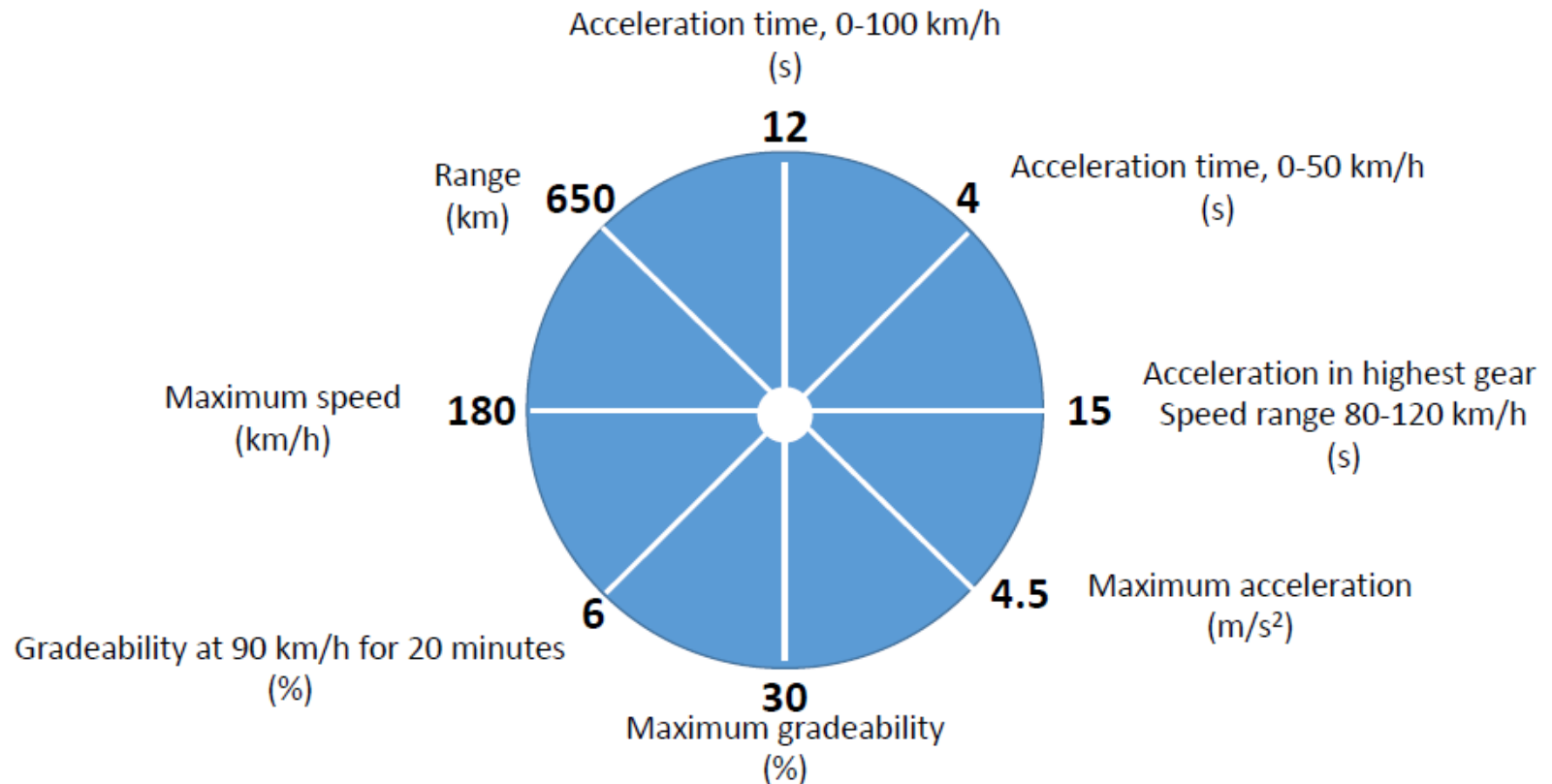




Fuel Cell Vehicles



Passenger Vehicle: Minimal customer performance criteria



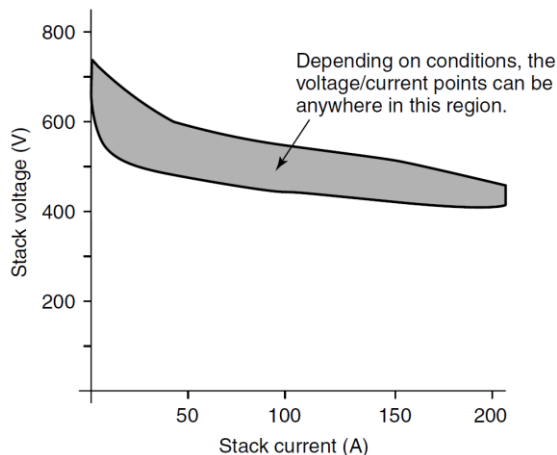
Fuel cell drivetrain

•FC System:

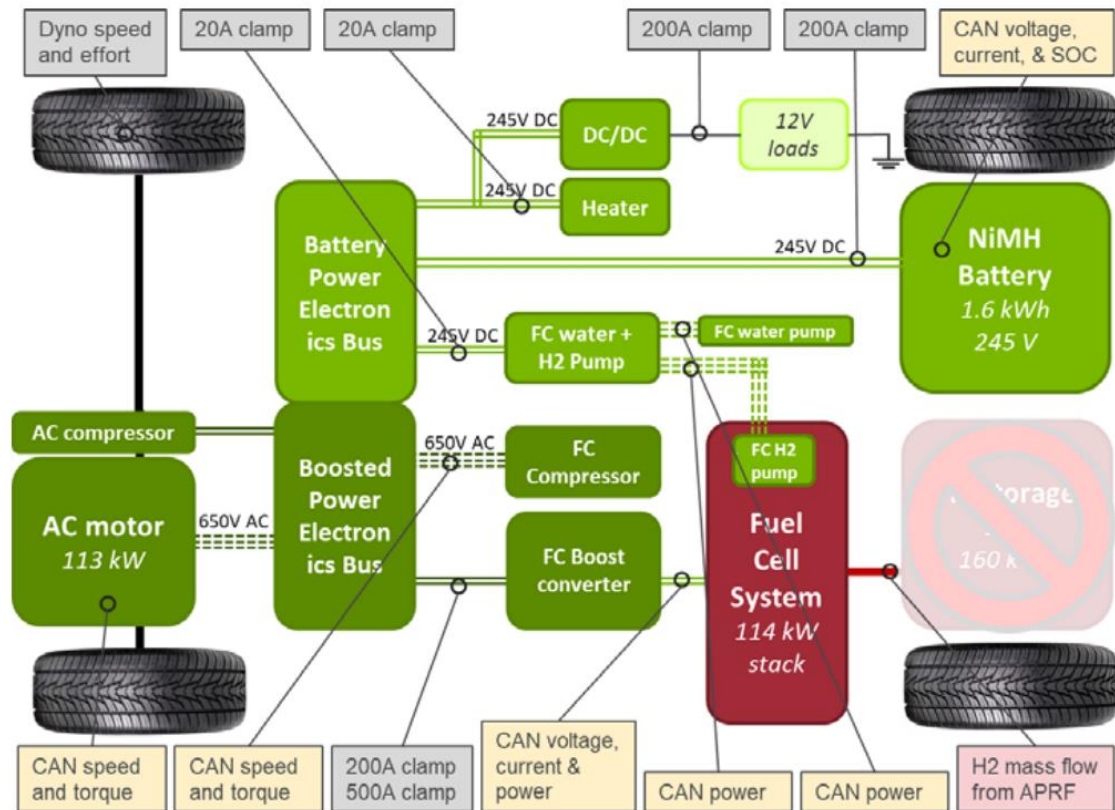
- Solid polymer electrolyte
- 370 cells in stack
- 114 kW, 3.1 kW/L, 2.0 kW/kg

•Battery:

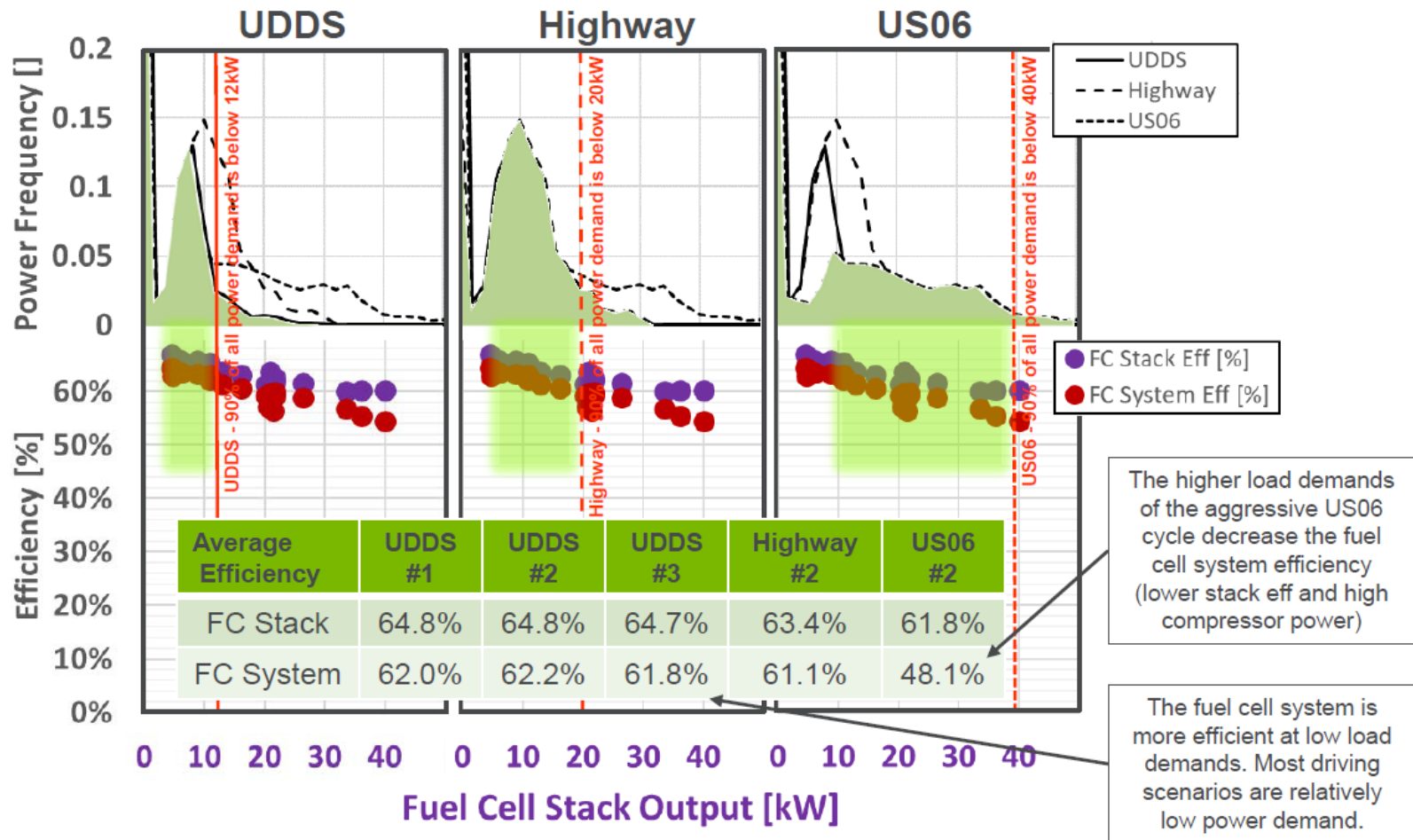
- Nickel-metal hydride
- 1.6 kWh, 245 V DC
- 34 cells
- Capacity 6.5 Ah



Toyota Mirai FCEV electric system (gen. 1)



Energy management - FC vehicles



Vehicle Force Analysis

The motion of the vehicle is opposed by a number of forces (resistances):

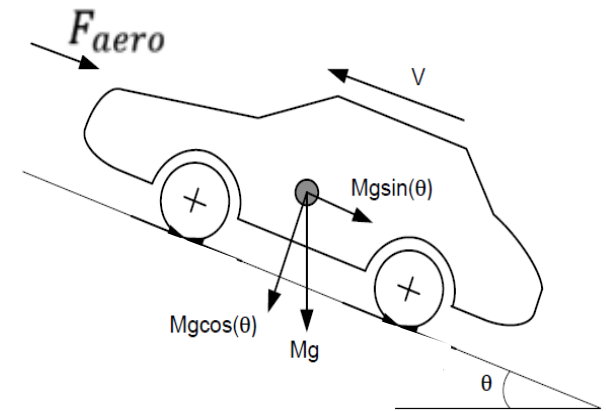
$$F = F_{aero} + F_{grade} + F_{rr} + F_i$$

Air drag (aerodynamic) resistance: $F_{aero} = \frac{1}{2} \rho C_d A_f v^2$

Road gradient resistance: $F_{grade} = m g \sin(\theta)$

Rolling resistance: $F_{rr} = m g C_{rr} \cos(\theta)$

Acceleration force: $F_i = m a$



C_d is the drag coefficient

ρ is the air density

g is the gravity acceleration (9.8 m/s^2)

a is the acceleration of the car

θ is the angle between the horizontal plane and the road

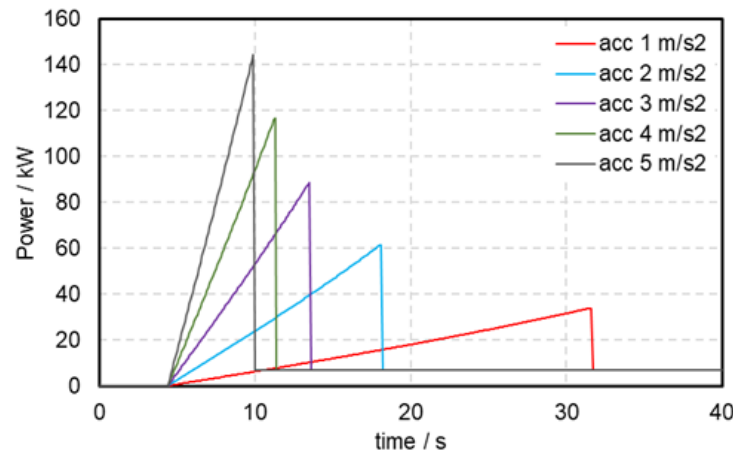
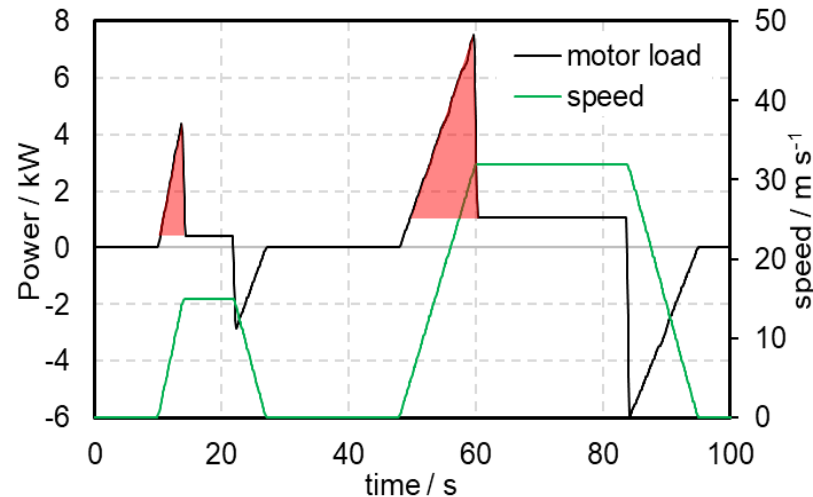
A_f is the cross-section area of the car

v is the car speed

m is the mass of the car

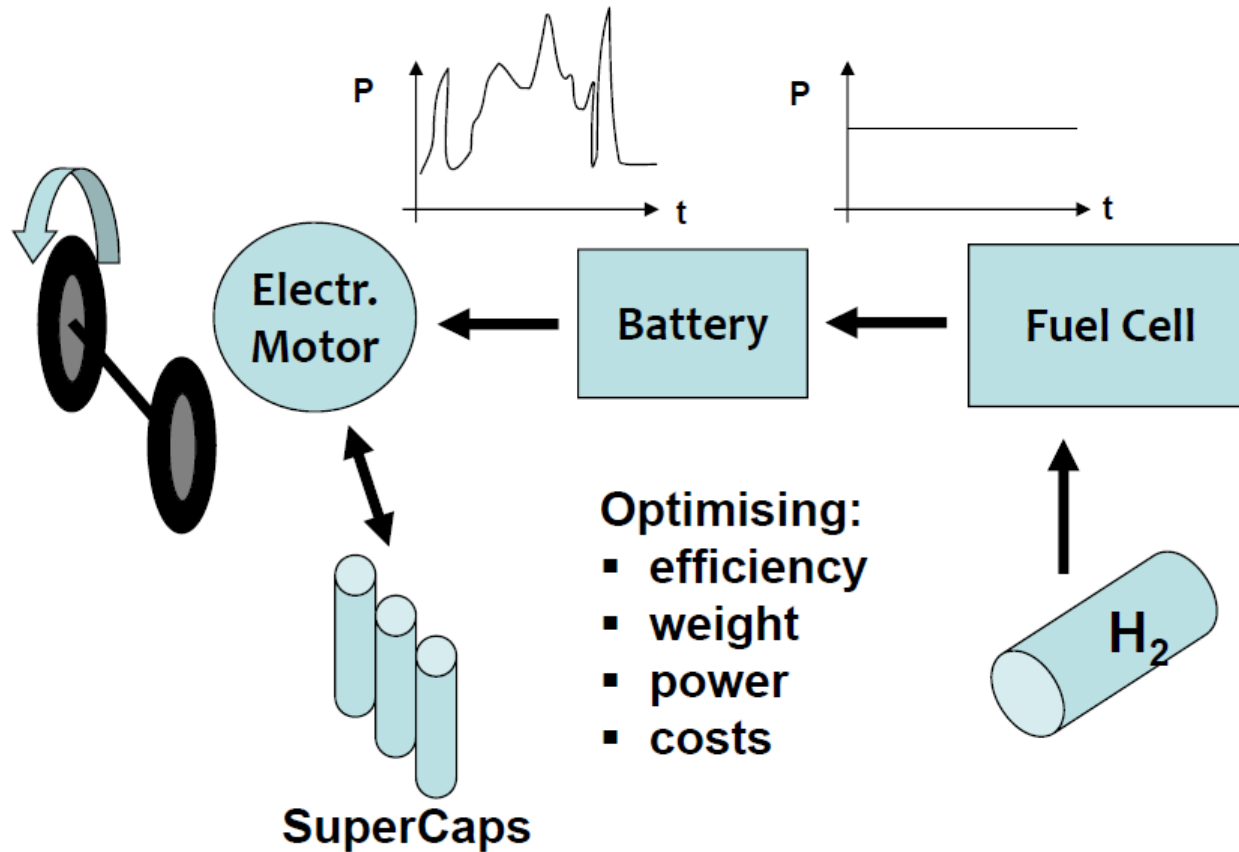
C_{rr} is the rolling friction coefficient

Drive Cycle Analysis

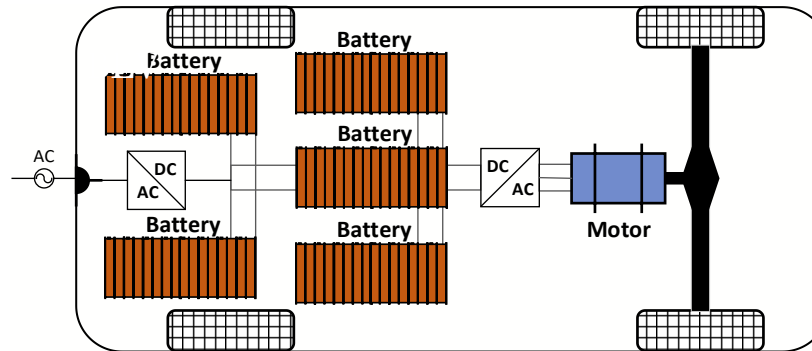


- At constant speed, acceleration is zero $\Rightarrow F_i = 0$
- Acceleration power demand represents a substantial amount of the vehicle's power requirement
- Driving style has a significant impact on the vehicle's power demand
- **Hybridising** can improve performance and efficiency

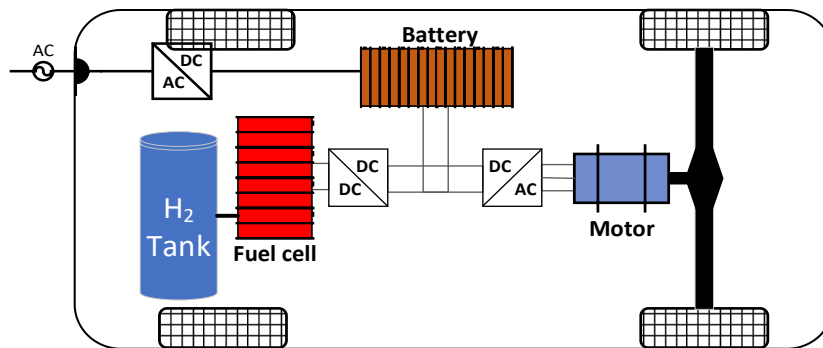
Hybridising



Hybridising



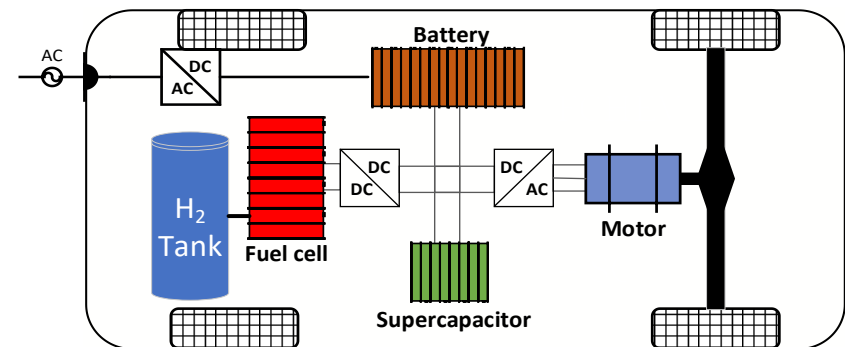
BEV



Plug-in option:

- Reduce frequent refuelling.
- Mitigate the need for expanding H₂ fuelling infrastructure

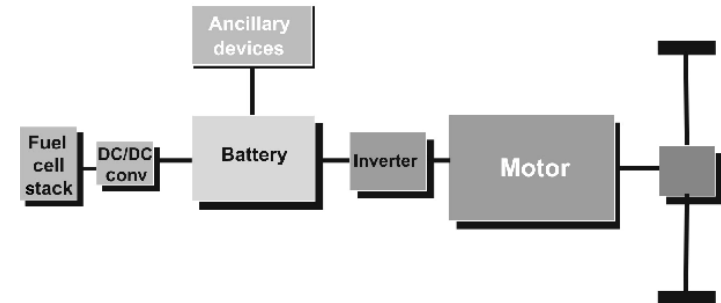
FCHEV



Series and Parallel FCHEV

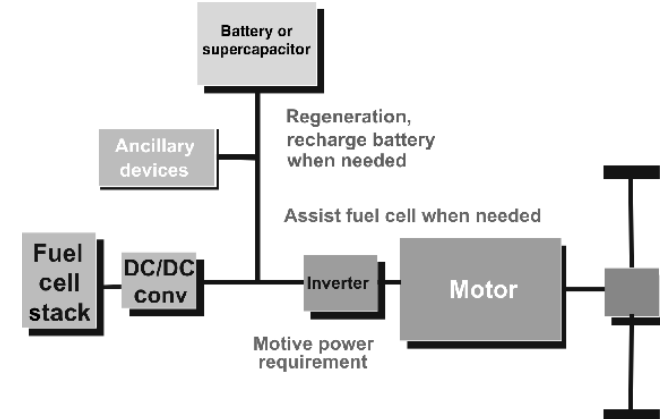
Series Hybrid:

- Small fuel cell, large battery bank
- Battery drives motor at all times
- Fuel cell operates continuously and keeps battery charged
- Fuel cell is not “load-following”



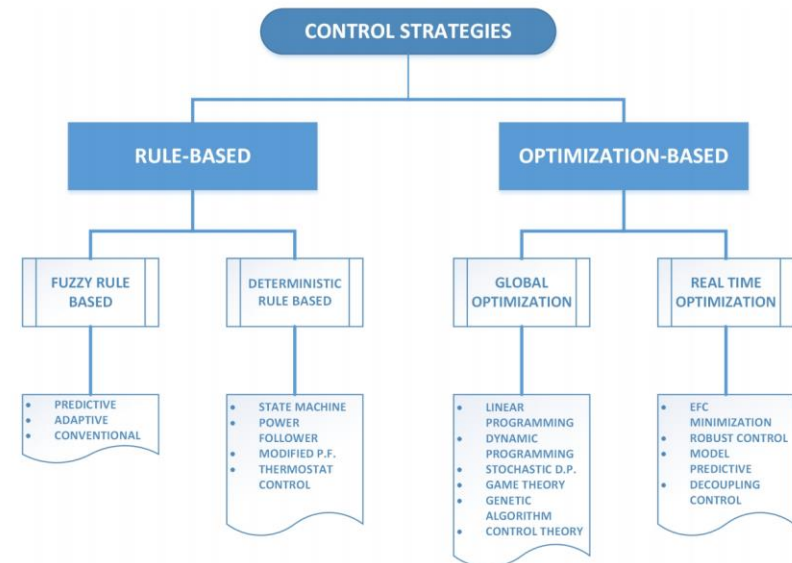
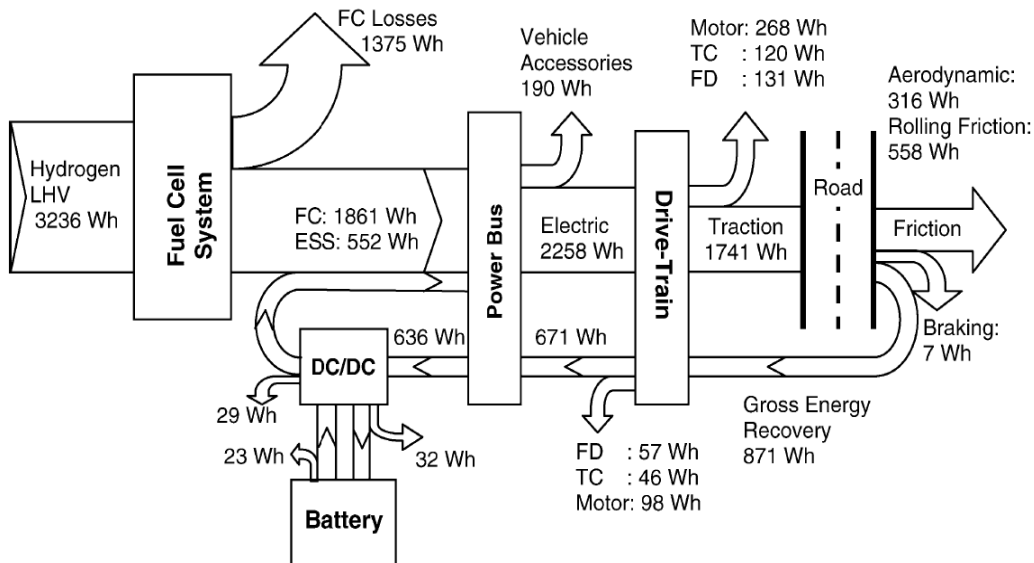
Parallel Hybrid:

- Large fuel cell, small battery bank
- Fuel cell drives motor most of the time “load-following”
- Battery provides boost power when required



Power Flow Management

Energy Flow within FCHEV drivetrain



Source: R.K. Ahluwalia

Off-road Vehicles

Agriculture machinery



Allis-Chalmers Fuel Cell Tractor 1959



Amogy's "ammonia to power"
zero-emissions tractor.



Harvesters

Off-road Vehicle: Fuel Cell Tractor

Differences between heavy goods and off-road vehicles

	HGV (trucks)	Off road vehicle
Working Speed	60 (miles/hr)	5 (miles/hr)
Cooling	Ram Air	Fan Power
Intake Air	Clean	Variable
Work Surface	Pavement	Dirt
Refill	At Station	At Machine
Usage	Continuous	Seasonal
Transient Demand	Medium	High
Off-Level	Minor	Substantial
Vibration	Mild	Severe
Environment	Clean	Debris



- (i) A Utility tractor, small farms, power range 50-100 kW, low energy usage.
- (ii) medium size tractor for dairy and livestock, a power range 100-200 kW, medium energy usage.
- (iii) A large size tractor for crop farming in big farms, power range 200-450+ kW, high energy usage.

JCB Fastrac 4220 Tractor

- Diesel engine efficiency $\sim 30\text{-}35\%$
- Tank capacity 390 L
- Rated power 217 hp = ~ 161.8 kW
- Maximum power 235 hp = ~ 175.2 kW
- Power Take Off 195 hp = 145.4 kW



JCB Fastrac 4220 Tractor

- The calorific value of diesel fuel 36.9 MJ/L (~ 10 kWh/L)
- Energy equivalent (diesel) $10 \text{ kWh/L} \times 390 \text{ L} = 3900 \text{ kWh}$
- Energy equivalent (H₂) $3900 \times 0.3 / 0.5 = 2340 \text{ kWh}$
- H₂ tank capacity = $2340 \text{ kW} / 39 \text{ kWh/kg} = 60 \text{ kg}$
- @ 350 bar ($\sim 24 \text{ kg/m}^3$): 2.22 m^3 (5.7x diesel tank)
- @ 700 bar ($\sim 42 \text{ kg/m}^3$): 1.23 m^3 (2.7x diesel tank)



Hydrogen Storage System

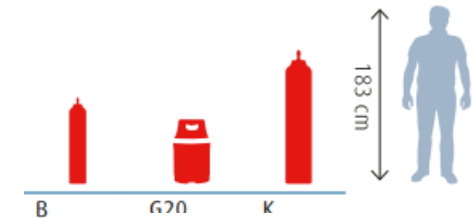
Type 1 Gas Cylinders – All Metal

Type 2 Gas Cylinders – hoop-wrapped aluminium

Type 3 Gas Cylinders – Fully-wrapped carbon composite

Type 4 Gas Cylinders – Full composite

Type 5 Gas Cylinders – Linerless Composite

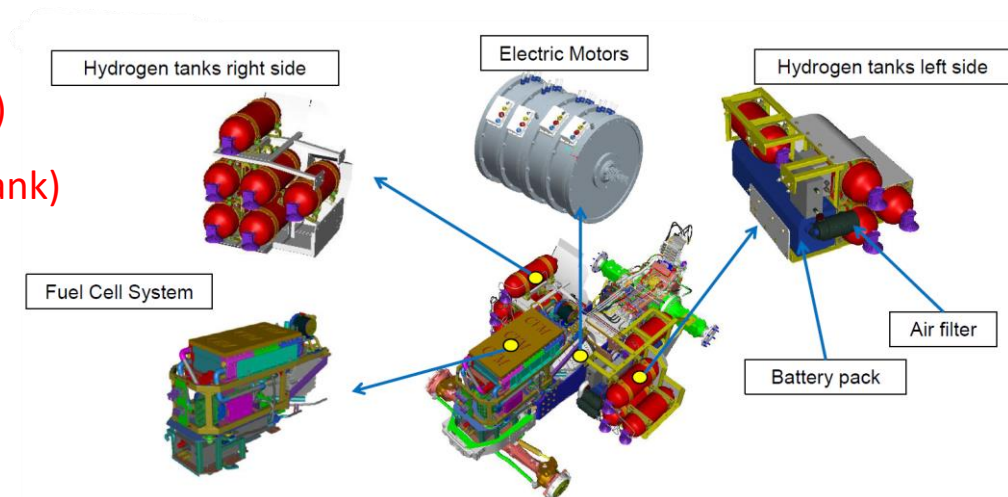


	H2 content kg	Gross weight (kg)	wt%	Bottle pressure bar	H2 density kg/m3	External Volume (L)	Internal volume (L)
K	0.585986	65	0.901516	175	12.70	61	46.123
B	0.120286	19	0.633082	175	12.70	13	9.46769
G20	0.404745	26	1.55671	300	20.07	53	20.15911
Mirai @10-15C	4.9878	92.49306	5.7	700	40.75	165.24	122.4
Luxfer (UK)	7.7	148.7	5.178211	350	24.23	431.5	322

Ratio (External/Internal) ~ 1.35

JCB Fastrac 4220 Tractor

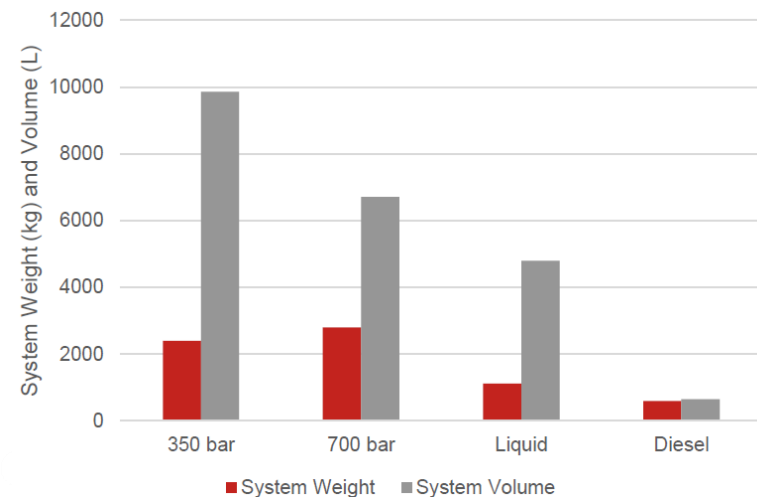
- Tank internal volume
 - @ 350 bar (26.1 kg/m^3): 2.3 m^3 (5.8X diesel tank)
 - @ 700 bar (42 kg/m^3): 1.23 m^3 (2.7X diesel tank)
- Tank system volume (+35%)
 - @ 350 bar: 3.1 m^3 (8x diesel tank)
 - @ 700 bar: 1.66 m^3 (4.5x diesel tank)



CNH Industrial

Challenges (FC Tractor)

- Cost
 - The total cost of fuel cell drivetrain is currently **15X** more expensive than a diesel powertrain
 - The DOE report: the cost fuel cell system and storage can be in parity with diesel if we factor in mass production of fuel cell drivetrain.
- Fuel cell power density:
 - Volumetric power density of fuel cell system includes the electric motor and inverter is ~ 170 kW/m³, while it is around 280 kW/m³ for diesel powertrains.
- H₂ fuel density:
 - A large size tractor of 650 L of diesel
 - 2 tones heavier with H₂ fuel
 - H₂ tank volume is an issue
- Stack cooling:
 - A huge radiator is required



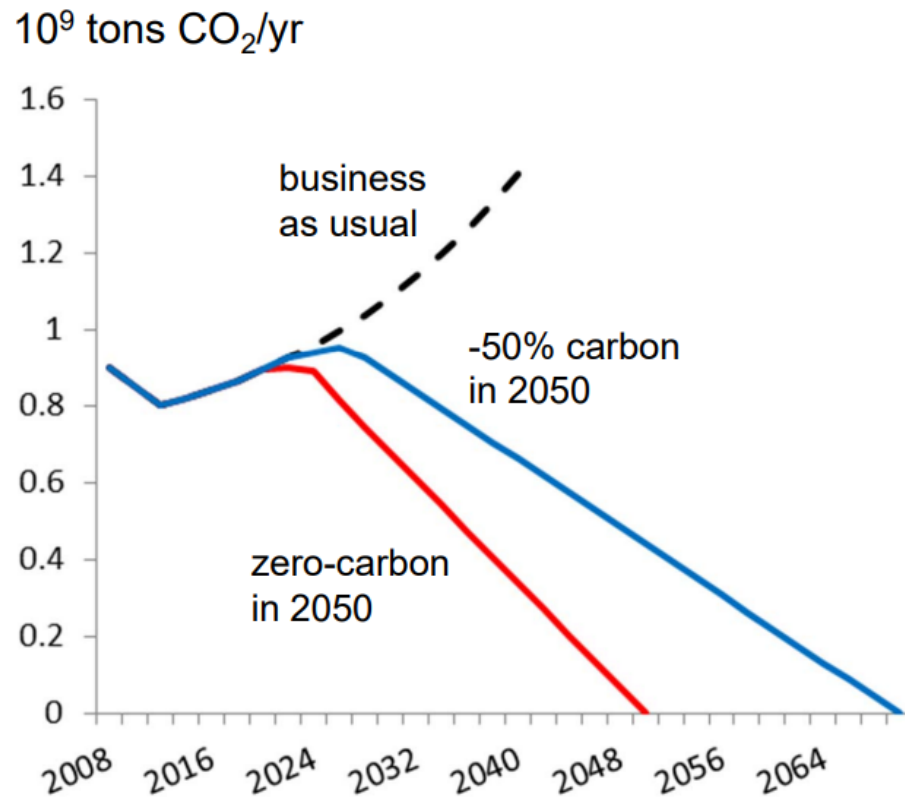
Weight and volume requirements of hydrogen system for a tractor.

CNH Industrial

Fuel Cells in Maritime Applications



Emission Reduction Scenario of the IMO



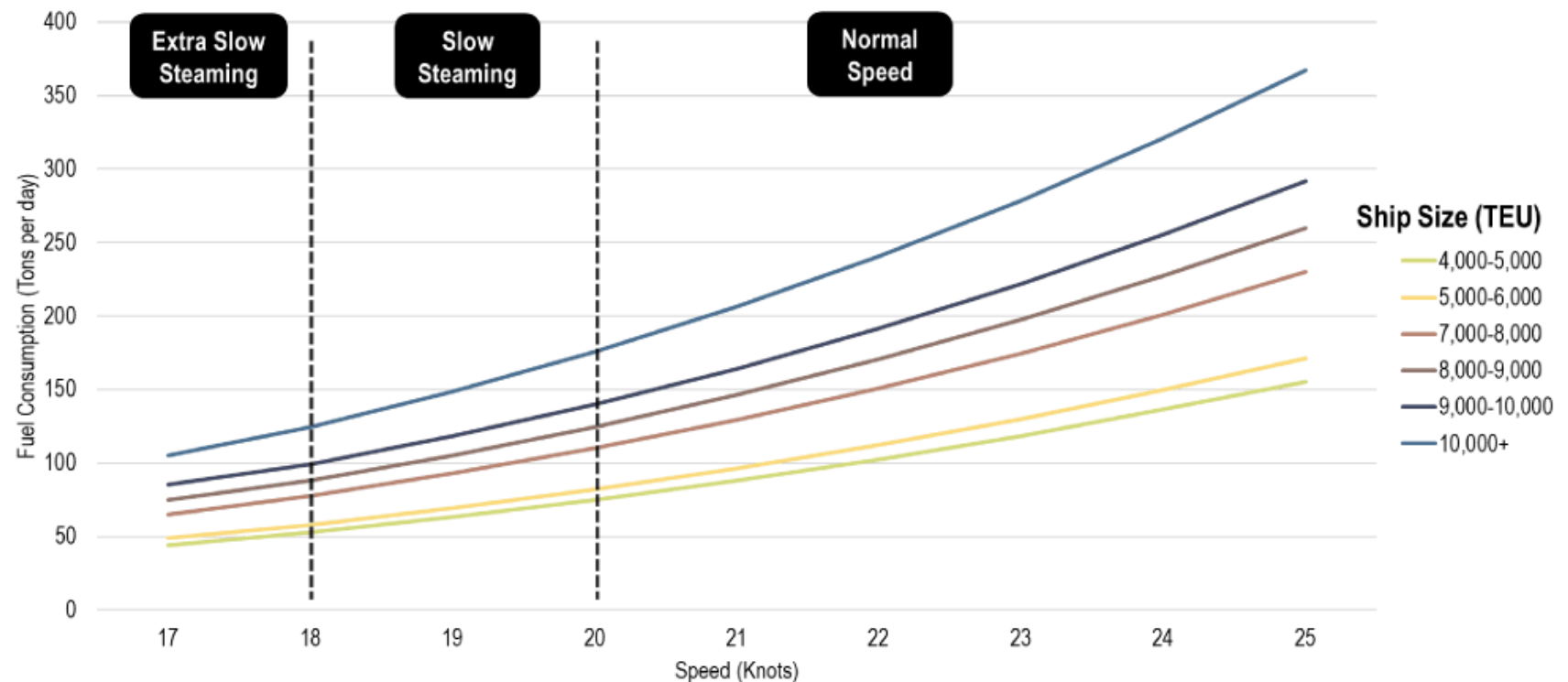
so what now?

- LNG?
- hydrogen?
- methanol?
- ammonia?
- methane?

What Amounts of Fuel are we talking about?

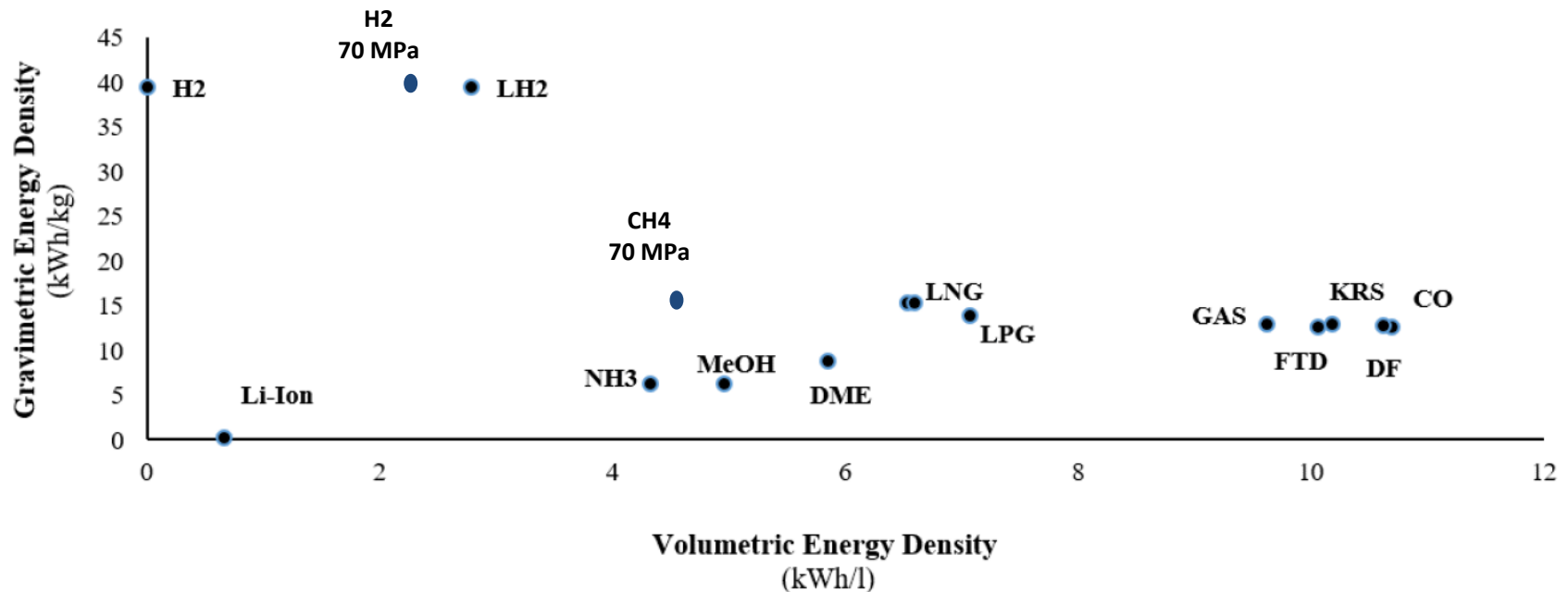
- For shipping

- Steaming distance 10 days, 150 to/dy, 16 500 MWh



source: transportgeography.org

Energy Density Comparisons



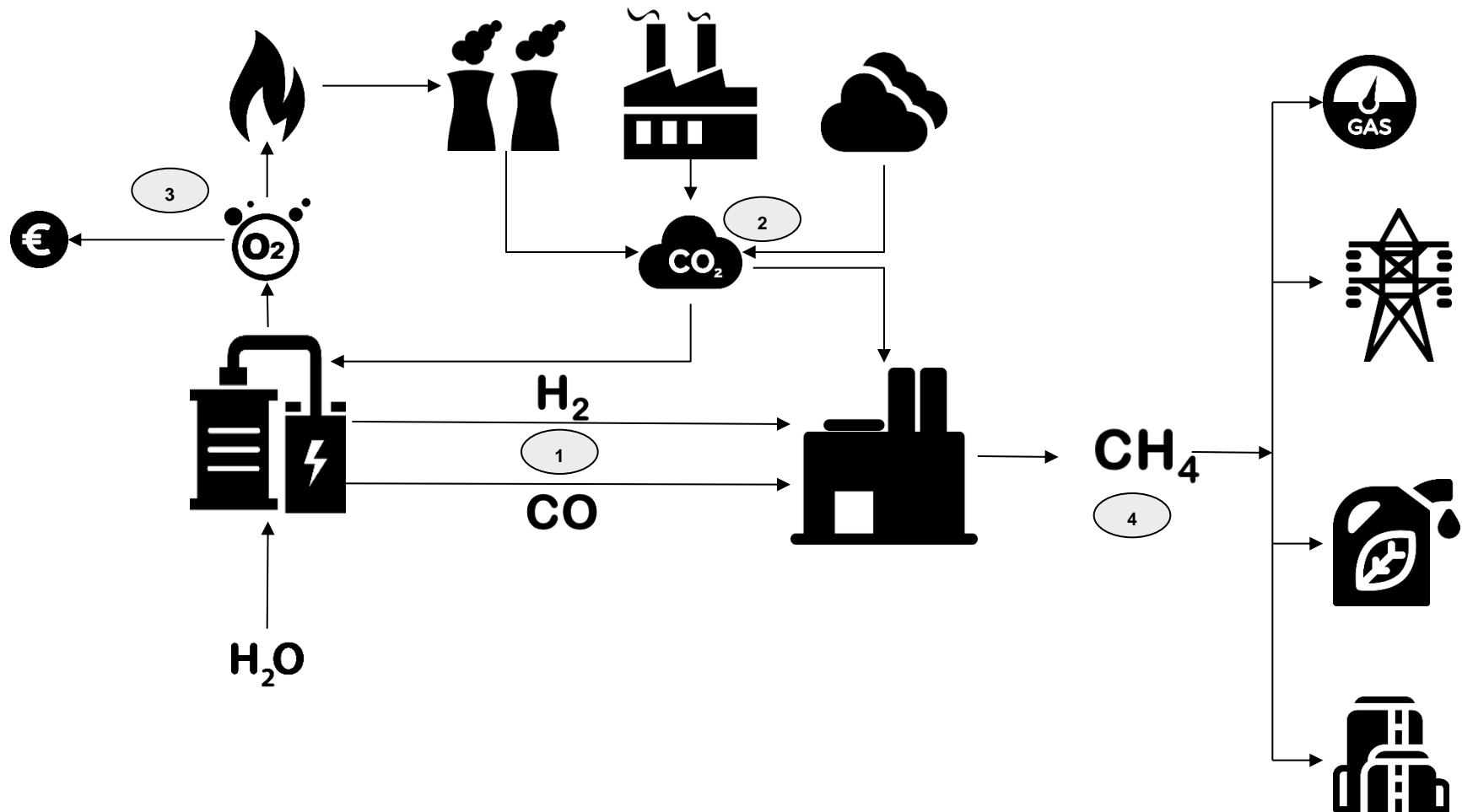
Where: GAS is gasoline, KRS is kerosene, DF is diesel fuel and CO is crude oil.

Comparison: Fuel Weight & Volume

- Steaming distance 10 days, 150 to/day, 16 500 MWh

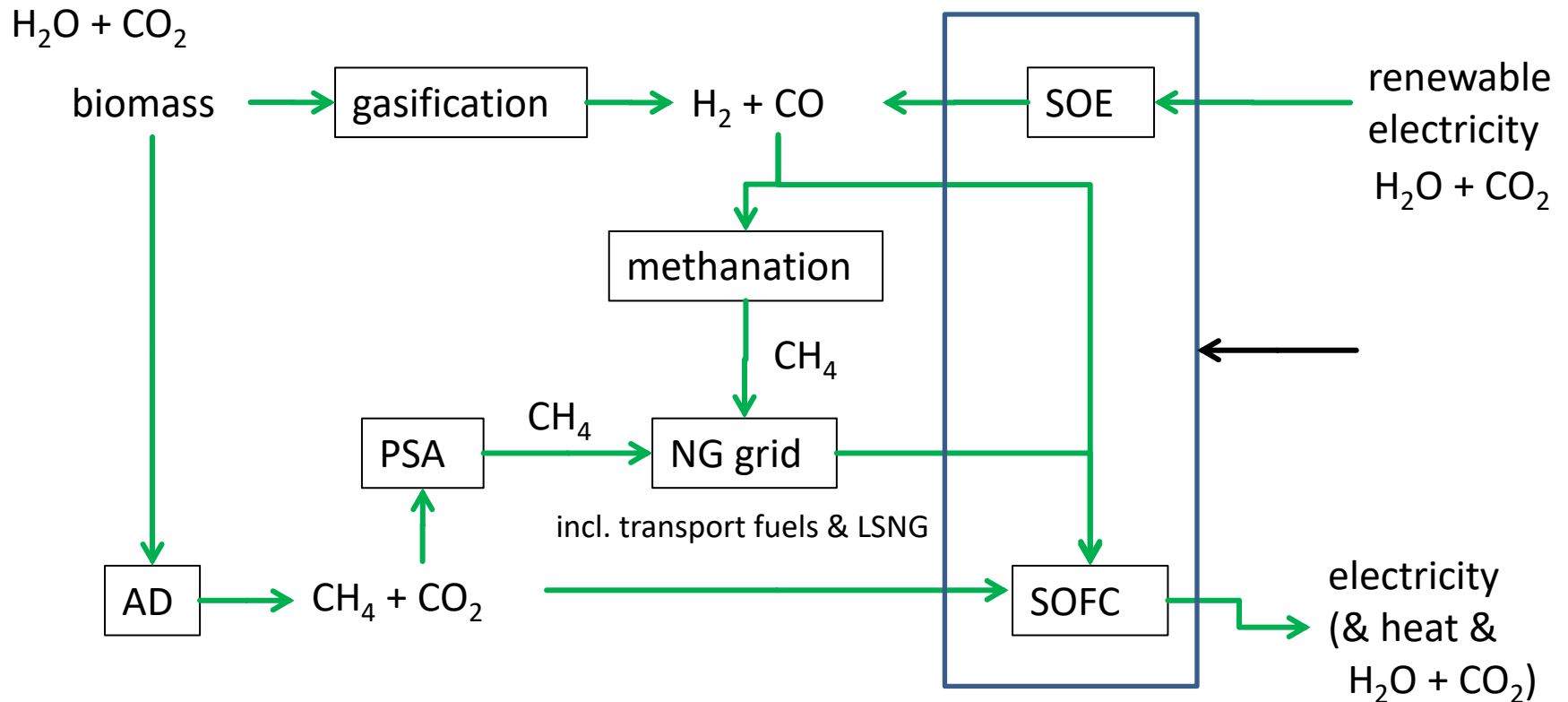
	Boil. Temp. [K]	Heat.Val. [kWh/kg]	Heat.Val. [kWh/L]	Heat.Val. [kWh/ Ncbm]	bunker- volume [cbm]	Fuel weight [to]	
LH2	20	33	2.3	---	7000	500	
LSNG	111	14	6.3	---	3200	1200	
LNG	120	12.5	5.6	---	3500	1300	
H2	---	33	---	3	9900	500	70 MPa
CH4	---	14	---	9	8900	1300	25 MPa
NH3	240	5.2	3.2	3.9	5200	3350	
CH3OH	---	5.5	4.4	---	3750	3250	
M.Diesel	---	12	10	---	1500	1500	

Power to Gas from CCU



From: Samuel Sogbesan/Robert Steinberger-Wilckens

The Cycle of Zero-Carbon Methane



supplies synthetic natural gas for stationary applications and transport fuels without any fossil carbon conversion involved

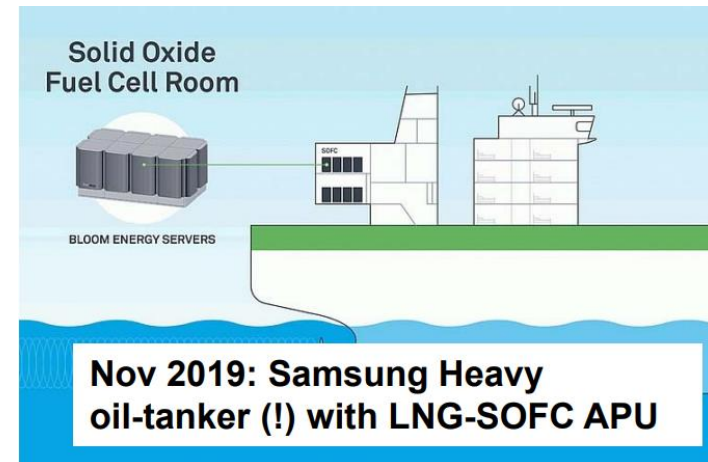
Zero-Net Carbon Methane (CH_4) for electrifying large vehicles

- fully compatible with natural gas (NG) grid infrastructure
- compatible of biomass-based fuels with NG/LNG trend
- hybridise vehicles with existing NG engine technology and Solid Oxide Fuel Cells (SOFC)
- no fossil carbon emission with massive reductions in CO, particle and noise emissions



Outlook

- besides the fuels, efficiency and emissions from shipping main engines play a major role in the emissions management in harbours
- electric propulsion based on fuel cells is a key technology to tackle harbour emissions
- today's decisions should already prepare for this pathway
- the choice of fuel should be based on the complete picture of emissions and energy use





Educational Activities

- KnowHy – distance / e-learning technician training: , L5 Train-the-Trainer programme.
- TeachHy – European MSc course in FCH technologies – delivered in blended learning (25% face2face and 75% e-learning)
- Centre for Doctoral Training in Fuel Cells and their Fuels.. (mini CDT programme 2022-23)
- Joint European Summer School on Fuel Cell, Electrolyser, and Battery Technologies (JESS)
<https://www.jess-summerschool.eu/>
- we are building a ,portal‘ as entry point for the above and adding general public suitable material such as MOOC‘s, general information etc.
- HyAcademy.EU Project 2024-2028





MSc / MRes / CPD modules

Section	No.	Title
Core / Mandatory (for MSc)	C1	Introduction to Electrochemistry
	C2	Fuel Cell Technologies and Applications
	C3	Hydrogen and hydrogen-based fuels
	C4	Fuel cell modelling tools and control
	C5	Characterisation methods
	C6	Fuel Cell and Hydrogen Lab
	C7	(Principles of) Hydrogen Safety
Optional / Elected	O1	Environmental analysis, LCA
	O2/O3	Low/High temperature fuel cells
	O4	Fuel Cell Systems
	O5	Advanced electrochemical applications
	O6	Fuel cell electric vehicles
	O7 / O8	Electrolysers and Hydrogen Infrastructure
	O9	Hydrogen markets, policies, RCS
	O10	Energy systems and Energy storage
	O11	Advanced modelling

MSc: 180 credit
120 taught module
60 research project

MRes: 180 credit
60 taught module
120 research project



www.hydex.ac.uk

Thank you